

1	1.	(Amended) A method for recovering material from waste tires, the method comprising the	
2		steps o	of:
3		(a)	contacting substantially whole waste tires with a molten reactant metal for a
4			reaction period;
5		(b)	collecting process gases released from the molten reactant metal during the
6			reaction period, the collecting of process gases including positioning a gas
7	\cap		recovery hood in an operating position in which a lower edge thereof extends into
8	M		the molten reactant metal around the area in which the waste tires contact the
9	O		molten reactant metal, the gas recovery hood defining a gas collection area in
10			which the process gases released from the molten reactant metal are trapped;
11		(c)	containing the waste tires on a tire carrier when the waste tires are contacted by
12			the molten reactant metal; and
13		(d)	removing unreacted solids from the molten reactant metal after the reaction period,
14			the unreacted solids being contained on the tire carrier for removal.
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16	6.	(Amen	nded) The method of Claim 1 further comprising the step of:
17		(a)	directing process gases trapped in the gas collection area to an aqueous scrubber
18			and removing metal salts and carbon from the process gases with the aqueous
19	117		scrubber.
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21	7 .	(Amen	ded) The method of Claim 1 further comprising the step of:
22		(a)	purging the gas collection area of air prior to collecting substantial amounts of
23			process gases in the gas collection area.
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contacting a tire portion with a mohen reactant metal including aluminum for a reaction period sufficient to allow substantially all organic materials originally

included in the tire portion to react with the molten reactant metal;

(New) A method for recovering material from a tire, the method comprising the steps of

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(b) containing the tire portion on a tire carrier when the tire portion is contacted by the molten reactant metal; and

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removing the tire carrier and unreacted solids retained on the tire carrier from the molten reactant metal immediately after the reaction period, the unreacted solids comprising solids remaining after the tire portion has contacted the molten reactant metal for the reaction period.

9. (New) The method of Claim 8 further comprising the step of maintaining the temperature of the molten reactant metal at a minimum of approximately 800 degrees Celsius during the reaction period.

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10. (New) The method of Claim 8 wherein the steps of contacting the tire portion with the molten reactant metal and containing the tire portion on the tire carrier include:

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(a) lowering the tire portion into the molten reactant metal on the tire carrier; and

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(b) pressing the tire portion into the molten reactant metal with a tire contactor member extending across an area above the tire carrier.

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11. (New) The method of Claim 10 wherein the step of removing unreacted solids from the molten reactant metal includes:

- (a) lifting the tire contactor member and the tire carrier from the molten reactant metal and allowing the molten reactant metal to drain from around the unreacted solids, tire contactor member, and tire carrier; and
- (b) cooling the tire carrier and unreacted solids located on the tire carrier.

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12. (New) A method for recovering materials from a tire portion including stainless steel, the method comprising the steps of:

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- (a) immersing the tire portion in a molten reactant metal including aluminum or an aluminum alloy for a reaction period sufficient to allow substantially all organic materials originally included in the tire portion to react with the molten reactant metal, the molten reactant metal being held at a temperature at which stainless steel dissolves therein;
- (b) containing the tire portion on a tire carrier when the tire portion is immersed in the molten reactant metal; and
- (c) removing the tire carrier and unreacted solids retained on the tire carrier from the molten reactant metal upon completion of the reaction period, the unreacted solids including solids remaining after the tire portion has contacted the molten reactant metal for the reaction period.

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